Sample Topics for Entrance Examinations

Faculty of Veterinary Medicine University of Warmia and Mazury in Olsztyn

BIOLOGY

- 1. Describe structure of cell nucleus
- 2. Characterise organization of cytoplasm in mammalian cells (describe organelles)
- 3. Describe mitosis
- 4. Shortly characterise parts of the digestive system in mammals
- 5. Shortly characterise parts of the respiratory system in mammals
- 6. Shortly characterise parts of the urinary system in mammals
- 7. Shortly characterise parts of the female reproductive system in mammals
- 8. Shortly characterise parts of the male reproductive system in mammals
- 9. Describe structure and function of the heart in mammals
- 10. Characterise composition and functions of blood in mammals
- 11. In mammals, there are several types of glands. Please list them and briefly describe their functions
- 12. Please list the elements of adaptation of birds to active flight
- 13. Please describe briefly what is double respiration in birds
- 14. Briefly describe the function of the placenta
- 15. Briefly describe the function of the thymus
- 16. Please list the types of mammalian teeth
- 17. Name the characteristic that makes mammalian erythrocytes different from bird erythrocytes?
- 18. Please list the functions of the skin in mammals
- 19. Please list the function of feathers in birds
- 20. Please name the difference between an eukaryotic and a prokaryotic cell

CHEMISTRY

- 1. Basic terminology used in general and organic chemistry
- 2. The most common elements and functional groups used in organic chemistry
- 3. Names and symbols of relatively common periodic table elements as well as polyatomic ions
- 4. Periodic trends in general chemistry electronegativity, atomic size, ionization energies
- 5. The Octet rule Valence electrons and their tendencies to react
- 6. Acid / Base chemistry pH, pKa, pKb
- 7. Common types of reactions in general chemistry
- 8. Common types of reactions in organic chemistry
- 9. Reduction / Oxidation reactions
- 10. Bonding and inter / intra molecular forces
- 11. Basic physical chemistry specific heat, spontaneity, Gibbs free energy, endothermic versus exothermic reactions, reaction kinetics